

Modification of brewer's spent grain after sc-CO₂ extraction: improvement of sugar and phenolic compounds release

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BSG is the most abundant brewing industry byproduct (85%), generated after the mashing and wort filtration process.



VE-1



Carbohydrate composition of the BSG and the sc-CO₂ raffinate in a dry and free-fat basis (g/100 g $_{\rm BSG}$).

Carbohydrate	BSG (%)	Raffinate (%)
Glucans	42 ± 2	43 ±1
Xylans	15 ± 1	15.2 ± 0.1
Arabinans	8 ± 1	7.5 ± 0.5

CH's composition were not significantly different after sc-CO₂ treatment.



Supercritical CO, extraction 80 °C, 40 MPa

Double effect of sc-CO₂ in a biorefinery context to include the BSG into a circular economy concept:

✓ Green solvent for **oil recovery**

P-1

Pretreatment agent for further improvement of the enzymatic hydrolysis yield of the sc-CO₂ treated BSG

sc-CO₂ treatment enhanced glucose yields for all the enzyme concentrations assayed

Cellulose dose (% w/w)	Increase in glucose yield (%)
0.25	8
0.5	14



Enzymatic hydrolysis by cellulase



1,4-(1,3:1,4)-β-D-Glucan 4-glucanohydrolase, EC 3.2.1.4
from Aspergillus niger (Sigma-Aldrich)
Cellulase activity: 1.18 U/mg

Operating conditions

T= 50 °C pH= 5 (acetate buffer) 5% dry BSG (% *w/v*) % Cellulase = 0.25 % -1 %, enzyme:BSG ratio (*w/w*)

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The **improvement** in enzymatic hydrolysis rate and yield after sc-CO₂ treatment could be attributed to:

- > the **removal** of the **lipid** fraction.
- surface morphology modification.

Improvement of phenolic compounds release after sc-CO₂ treatment

Phenolic compounds release yield by different treatments

Treatment	Cumaric acid, µg/g _{BSG}	Vanillin, µg/g _{BSG}	Ferulic acid, µg/g _{BSG}
Celullase, 1 %	3.0 ± 0.3	20 ± 1	274 ± 4
sc-CO ₂ + Celullase, 1 %	3.9 ± 0.3	21 ± 2	341 ± 6



SEM micrographs (1000 x magnifications) of the different BSG samples: untreated BSG, sc-CO₂ treated BSG and after enzymatic hydrolysis

- Untreated BSG: more rigid and continuous surface
- sc-CO₂ treated BSG: irregular porosity and lamellar structure.



Xilanase, 1 %	6 ± 1	111 ± 3	52.4 ± 0.9
Alakaline hydrolysis	538 ± 4	217 ± 1	1305.7 ± 0.5
Subcritical water 185 °C	60 ± 8	330 ± 11	144 ± 10

- ✓ p-Coumaric acid concentration increased 30%
- ✓ Ferulic acid concentration increased 25%
- ✓ The concentration of vanillin was similar in both hydrolysates.

The release of ferulic acid by CO_2 + Celullase, 1 % was noticeable higher than those obtained by other hydrolytic methods with exception than alkaline hydrolysis

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