

¹ Fluorescent Discrimination between Traces of Chemical Warfare ² Agents and Their Mimics

³ Borja Díaz de Greñu,[†] Daniel Moreno,[†] Tomás Torroba,^{*,†} Alexander Berg,[‡] Johan Gunnars,[‡]
⁴ Tobias Nilsson,[‡] Rasmus Nyman,[‡] Milton Persson,[‡] Johannes Pettersson,[‡] Ida Eklind,[§]
⁵ and Pär Wästerby^{*,§}

6 [†]Department of Chemistry, Faculty of Sciences, University of Burgos, 09001 Burgos, Spain

7 [‡]Department of Physics, Umeå University, SE-90187 Umeå, Sweden

⁸ [§]Department for CBRN Defense and Security, Swedish Defense Research Agency (FOI), SE-90182 Umeå, Sweden

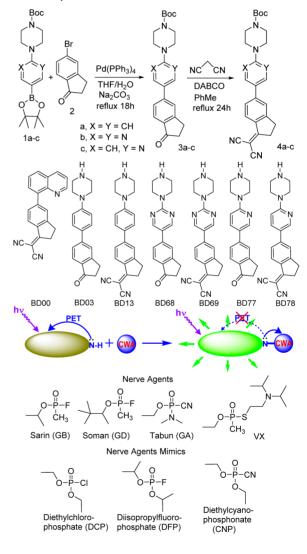
9 Supporting Information

ABSTRACT: An array of fluorogenic probes is able to 10 discriminate between nerve agents, sarin, soman, tabun, 11 VX and their mimics, in water or organic solvent, by 12 qualitative fluorescence patterns and quantitative multi-13 variate analysis, thus making the system suitable for the in-14 the-field detection of traces of chemical warfare agents as 15 well as to differentiate between the real nerve agents and 16 other related compounds. 17

Terve agents are highly toxic volatile liquids that 18 irreversibly block the enzyme acetylcolinesterase in the 19 20 neuronal synapsis, thus disrupting nerve impulse transmission 21 and causing death through the paralysis of respiratory muscles.¹ 22 They are used as chemical warfare agents (CWA) for dirty war 23 in undeveloped countries, causing hundreds of victims, 24 although their use as chemical weapons is prohibited.² Their 25 quick detection can be achieved by hand-held instruments that ²⁶ are costly and prone to false positives³ so the availability of safe 27 and easy to use portable devices is most sought-after. More 28 importantly, the investigation of chemical weapons allegations 29 is a very slow process that implies unequivocal detection of 30 CWA residuals in water and organic samples,⁴ with the risk of 31 long delays in the environment of worrying war scenarios.⁵ 32 Colorimetric⁶ or fluorimetric⁷ reactive dyes in solution or as 33 arrays,⁸ as well as supported in nanomaterials,⁹ have been used 34 for fast detection of CWA as good alternatives to classic 35 methods, but most of these methods are implemented for nerve 36 agents mimics, and so there is no clear proof that they will work 37 for real CWA.¹⁰ To complement the existing methodologies, 38 we have developed a series of new highly solvatochromic 39 fluorescent indicators for phosphorylating reagents capable of 40 developing large differences in fluorescence. In this paper, we 41 report our findings upon the selective fluorescent discrim-42 ination of real nerve agents from their mimics.

⁴³ We have previously prepared some charge-transfer fluoro-⁴⁴ genic probes, bearing conjugated donor and acceptor groups in ⁴⁵ their structure, that were useful for the detection of significant ⁴⁶ analytes.¹¹ For our current purpose we have designed new ⁴⁷ fluorescent probes (Scheme 1).

⁴⁸ In this case, they have a secondary donor group that was not ⁴⁹ involved in the charge-transfer process. Thus, the Suzuki



Scheme 1. Synthesis of Fluorescent Probes and Their Action

Received: January 22, 2014

s1

50 reaction of aryl boronates 1a-c and 5-bromoindanone 2 $_{51}$ catalyzed by Pd(PPh₃)₄ in tetrahydrofuran/water in the 52 presence of Na₂CO₃ gave arylindanones 3a-c in 85-95% 53 yields. Knoevenagel reaction of 3a-c and malononitrile in the 54 presence of DABCO in toluene at reflux for 24 h gave 55 arylindanes 4a-c in 55-68% yields. N-Boc deprotection with 56 trifluoroacetic acid for 15 min from 3a-c and 4a-c gave the 57 unprotected amine derivatives BD03, BD13, BD68, BD69, 58 BD77 and BD78, in which the initial fluorescence of 3a-c and 59 4a-c is quenched in some extension by a photoinduced 60 electron transfer from the free amine group. Subsequent 61 acylation or phosphorylation of the amine group should 62 therefore increase fluorescence of these compounds, thus 63 making these compounds suitable for phosphorylating agents 64 detection (Scheme 1). Fluorescence of these compounds can 65 be also affected by protic acids, therefore we added to the series 66 a fluorogenic dye, BD00,¹² which is not fluorescent but 67 develops a blue fluorescence in the presence of common protic 68 acids. In this way, false positives are prevented. We next tested $69 \, 10^{-4}$ M solutions of the seven fluorescent probes in 70 dimethylsulfoxide (DMSO) or acetonitrile (MeCN) with 1 71 equiv of 5×10^{-3} M solutions of nerve agent simulants (DCP, 72 DFP, CNP) (Scheme 1) and phosgene¹³ (Cl₂CO) in MeCN or 73 water and recorded all changes that the fluorescent probes 74 underwent with every analyte under a common TLC-UV light, 75 λ = 366 nm, by qualitative (photographs) and quantitative 76 measurements such as initial and final λ_{\max}^{abs} and λ_{\max}^{fluo} , 77 variations in the relative intensity of fluorescence and kinetics of 78 processes. The qualitative measurements gave clear and distinct 79 fingerprints of every nerve agent mimic used for testing the 80 probes, undoubtedly discriminating between them. The 81 quantitative measurements were subjected to hierarchical 82 cluster analysis (HCA).⁸ HCA dendrogram obtained from 83 fluorescent measures showed a clear clustering for all the nerve 84 agent simulants, blank and phosgene, giving a good separation 85 of every analyte (Figure S65b). Absorbance or mixed data from 86 absorbance and fluorescence afforded a poor separation 87 between some analytes (Figures S64 and S65a), therefore 88 establishing that discrimination between analytes is best 89 obtained by fluorescence measurements. Likewise, principal 90 components analysis (PCA)¹⁴ of the same data afforded good 91 discrimination between each one of the CWA mimics as well as 92 phosgene (Figure S66), therefore probing that the array of 93 fluorescent dyes is able to discriminate between closely related 94 phosphorylating or acylating reagents by both their fingerprints, 95 HCA or PCA. The next step was testing the system with real 96 nerve agents, but because of the extreme toxicity we performed 97 the tests at the laboratories of the FOI CBRN Defense and 98 Security (Umeå, Sweden), where handling of nerve agents was 99 performed under appropriate conditions. Again, the seven 100 different fluorescent probes were mixed with a series of nerve 101 agents, Soman, Sarin, Tabun, and VX and chemically similar 102 substances diethylchlorophosphate (DCP) and diethylcyano-103 phosphonate (CNP), in the same conditions used for CWA 104 mimics. The acquired samples of mixtures were then subjected 105 to light (300-500 nm) in which they fluoresced with different 106 colors. Light intensities were registered with a spectrofluor-107 ometer and photographs were taken for a chart of visible colors 108 of all the test samples. The probes and CWA were solved in 109 two different solvents, DMSO and MeCN for the probes, and 110 MeCN and water for the CWA/CWA-simulants. The probes 111 were also tested without CWA or simulant. The acquired 112 mixtures were named as in the following example: Sarin solved

in water mixed with probe DM13 solved in DMSO was called 113 GB_W13D. For the mixtures with only probes the name begins 114 with NaN. To photograph the samples they were placed under 115 a 366 nm UV-lamp in a dark room. A color reference sheet 116 illuminated with white light was placed nearby (Figure S89). 117 Copies of the RAW-files were edited, all in the same way (batch 118 process), before being converted to JPG for extraction of the 119 colors as RGB-values. Both the colors from the edited images 120 and from the original images were analyzed. As an example, a 121 photograph of Soman samples is seen in Figure 1. 122 fil



Figure 1. Samples contained Soman in MeCN mixed with each of the seven probes in DMSO. From left to right the samples contained probes BD00 to BD78.

Since there were three images of every set of seven samples, 123 the mean values in R, G, and B had to be computed. A table of 124 these colors in the form of colored squares was then created as 125 seen in Figure 2. Looking at the tables of observed colors it was 126 f2

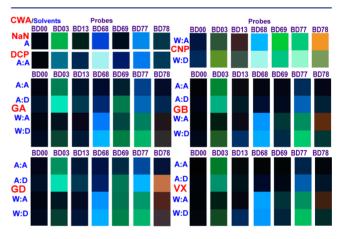


Figure 2. Observable colors in 366 nm excitation light. CWA: Sarin (GB), Soman (GD), Tabun (GA), VX (VX). CWA-simulants: CNP and DCP. CWA/simulants solvent: water (W) and acetonitrile (A). Probes: BD00, BD03, BD13, BD68, BD69, BD77, BD78. Probe solvents: dimethylsulfoxide (D) and acetonitrile (A).

clear that several probes could be used to guarantee the absence 127 of Sarin, Soman, Tabun, and VX. For some choices of solvents 128 there was a possibility to make the distinction between Soman 129 and the other CWA. If a sample with probe BD69 in 130 acetonitrile fluoresced very weakly (as in probe BD69 in 131 acetonitrile, known as NaN A in the table of observed colors), 132 then the risk of there being Sarin, Soman, Tabun, or VX in the 133 samples is low since the corresponding CWA samples, with 134 nerve agent in water and probe in acetonitrile, all fluoresced 135 green. Probe BD78 acts in a similar way, but here the CWA- 136 samples fluoresced in orange, while for probe BD03 it is the 137 other way around. The NaN A sample with probe BD03 138 fluoresced in bright green, while there is barely any fluorescence 139 from the corresponding CWA samples. Probe BD77 also gave 140 valuable information but in a different way. For this probe both 141 the probe and the CWA samples fluoresced clearly, but the 142

143 probe samples did so in blue, while the CWA samples all 144 fluoresced in green. It was probe BD78 that indicated that there 145 was a possibility to distinguish Soman from the rest of the 146 CWA. It is visible in the Figure 2 of observed colors that the 147 mixture of Soman in MeCN and probe in DMSO fluoresced in 148 a clear orange color, while the rest of the CWA samples with 149 the same solvents fluoresced with weak obscure blue color. A 150 table with the colors from the unedited images can also be 151 found in Figure S90. The colors from preliminary experiments 152 with only simulants have been included in a similar table in 153 Figure S92. For quantitative measurements we used a calibrated 154 spectrofluorometer. In the analysis of the spectral data a 155 multivariate data analysis with Simca¹⁵ software was used. To 156 analyze the data we used a couple of approaches. Some of the 157 basic analysis was made just by looking at the plots of the 158 spectroscopy data. We were able to see that some of the 159 mixtures just gave fluctuations in the data, while other gave 160 clear tops. We found that the probes BD03, BD68, and BD77 161 were the probes that gave the highest number of clear tops, 162 while the other only gave a few clear tops. In Figure 3 six

f4

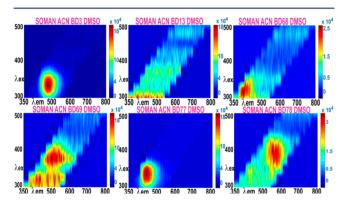


Figure 3. Plots of the spectrofluorometer data for CWA Soman (GD) solved in MeCN with the probes BD03 to BD78 in DMSO. In the plots we see how probe BD13 (top middle) causes fluctuations over the whole spectra while the probes BD03 and BD77 (top left and bottom middle) have very clear spectra. One of the few tops of probe BD78 (bottom right) can be seen and it is also this clear orange color that makes Soman stand out from the other CWA in the photographs.

163 spectra for Soman are plotted, showing all types of spectra that 164 occurred, along with the distinct spectra for probe BD78 that 165 distinguished Soman from the other probes in the spectra as 166 well as in the photographs.

By use of multivariate data analysis we found that we were 167 168 able to detect in which solvent the CWA were solved. We were 169 also able to see a clear difference between the probes that gave 170 clear tops in the spectrofluorometer data, and those that did 171 not. In the analysis of the spectrofluorometer data the measured 172 values were emission (λ_{em}) and excitation (λ_{ex}) wavelengths, and intensity of the maximum (fl max) in each of the produced 173 two-dimensional spectra. We also calculated the area in the 174 175 spectra with intensities of 50% and 75% or more of the 176 maximum (area50 and area75, respectively). We were able to see a clear difference between the simulants and the CWA 177 when performing a multivariate analysis on agent-probe 178 179 combinations (Figure 4). After our analysis we can conclude 180 that there is a large probability that the probes are able to detect 181 the most important CWA from their mimics. In the analysis 182 only probes BD03, BD68, and BD77 were used to avoid the 183 fluctuations, as variables we used area50, λ_{ext} , λ_{emt} , and fl max for

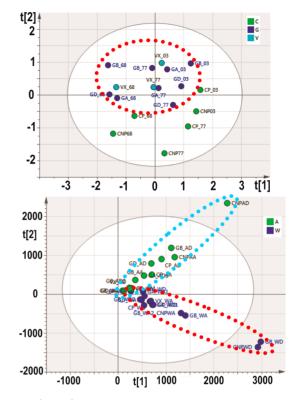


Figure 4. (Upper) A plot of the first two principal axes when running a multivariate analysis over the agent-probe combinations. In the plot we see a clear separation of the CWA (G and V) and the simulants (C). In this plot only the probes BD03, BD68, and BD77 were used, i.e., the probes that did not have a large tendency to fluctuate. (Down) A plot of the first two principal axes when running a multivariate analysis over the agent-solvent combinations. In the plot we can see a clear separation of the agents solved in water and those solved in acetonitrile.

each of the four solvent combinations. Before the PCA analysis 184 was run all data were transformed logarithmically and grouped 185 in blocks of $\lambda_{\rm em}/\lambda_{\rm ex}$, fl_max, and area50 before unit variance 186 was run. From the load vectors for the analysis of the agent- 187 probe combinations (Figure S72) we could see how the areas 188 and florescence were the main parameters for the second 189 component. In addition, the simulants generally had a bit 190 higher fluorescence and for some a bit smaller areas, therefore 191 they tended toward the lower values on the second component. 192 When studying the combinations of agent and solvents against 193 probes we were able to see a clear separation between the 194 agents that were solved in water and those that were solved in 195 acetonitrile. This can clearly be seen in Figure 4, the main 196 reason behind this separation seems to be that the intensity of 197 probe BD68 becomes higher for those agents solved in water 198 (Figure S88).

In summary, we have synthesized a new series of fluorogenic 200 probes that are able to discriminate between traces of CWA and 201 their mimics, in water or organic solvent. Discrimination is 202 achieved by means of the different fluorogenic response 203 triggered by CWA or their mimics on the fluorogenic probes 204 in different solvent combinations of CWA and probes. 205

The different response given by the series of fluorogenic 206 probes is charted as a fingerprint of the fluorescent response of 207 every CWA/probe/solvent combination under a common 366 208 nm UV light, thus permitting a fast visual differentiation 209 between CWA and their mimics. More accurate discrimination 210

211 is achieved by multivariate analysis by using quantitative 212 measurements in fluorescence spectroscopy. In this way we 213 have obtained a complete differentiation between CWA and 214 their mimics, so the system is suitable for the accurate in-the-215 field detection of traces of CWA. We have seen that the 216 response given by CWA mimics is very different to the 217 response given by the real CWA, because of the slightly 218 different chemical functionality of CWA and their mimics. 219 Since most of the chromogenic and fluorogenic probes hitherto 220 studied for the detection of CWA are based in the study of the 221 response given by their mimics, there is no guarantee that 222 previously known probes for CWA mimics will work with real 223 CWA samples. Our work clearly shows that the response can be 224 very different. In addition, the synthesis of the reported 225 fluorogenic probes is simple and straightforward, therefore 226 these fluorescent probes are suitable for the development of 227 upcoming practical methodology.

228 ASSOCIATED CONTENT

229 Supporting Information

230 Experimental details and characterization data. This material is 231 available free of charge via the Internet at http://pubs.acs.org.

232 **AUTHOR INFORMATION**

233 Corresponding Authors

234 ttorroba@ubu.es

235 par.wasterby@foi.se

236 Notes

237 The authors declare no competing financial interest.

238 **ACKNOWLEDGMENTS**

239 We gratefully acknowledge financial support from the Minister-240 io de Economía y Competitividad, Spain (Project CTQ2012-241 31611), Junta de Castilla y León, Consejería de Educación y 242 Cultura y Fondo Social Europeo (Project BU246A12-1), the 243 European Commission, Seventh Framework Programme 244 (Project SNIFFER FP7-SEC-2012-312411) and the Swedish 245 Ministry of Defence (no. A403913). This paper is dedicated to 246 the memory of Dr. Stefano Marcaccini who passed away 247 October 1, 2012.

248 **REFERENCES**

(1) Reviews: (a) Russell, A. J.; Berberich, J. A.; Drevon, G. F.;
Koepsel, R. R. Annu. Rev. Biomed. Eng. 2003, 5, 1. (b) Eubanks, L. M.;
Dickerson, T. J.; Janda, K. D. Chem. Soc. Rev. 2007, 36, 458.
(c) Ganesan, K.; Raza, S. K.; Vijayaraghavan, R. J. Pharm. BioAllied Sci.
2010, 2, 166. (d) Sambrook, M. R.; Notman, S. Chem. Soc. Rev. 2013,
42, 9251. See also: (e) Chemical Warfare Agents: Toxicology and
Treatment, 2nd ed.; Marrs, T. C., Maynard, R. L., Sidell, F., Eds.; John
Wiley & Sons: Chichester, 2007.

257 (2) Convention on the prohibition of the development, production, 258 stockpiling and use of chemical weapons and on their destruction, 259 http://www.opcw.org/chemical-weapons-convention/download-the-260 cwc/.

261 (3) Sferopoulos, R. A Review of Chemical Warfare Agent Detector 262 Technologies & Commercial-Off-The-Shelf Items; DSTO Defence 263 Science and Technology Organisation: Victoria, Australia, 2009.

264 (4) Review: (a) Aleksenko, S. S. J. Anal. Chem. 2012, 67, 82. E.g.:
265 (b) Cordell, R. L.; Willis, K. A.; Wyche, K. P.; Blake, R. S.; Ellis, A. M.;
266 Monks, P. S. Anal. Chem. 2007, 79, 8359. (c) Kolakowski, B. M.;
267 D'Agostino, P. A.; Chenier, C.; Mester, Z. Anal. Chem. 2007, 79, 8257.
268 (d) Nilles, J. M.; Connell, T. R.; Durst, H. D. Anal. Chem. 2009, 81,
269 6744. (e) Knaack, J. S.; Zhou, Y.; Magnuson, M.; Silvestri, E.; Johnson,
270 R. C. Anal. Chem. 2013, 85, 2611. (f) Esipenko, N. A.; Koutnik, P.;

Minami, T.; Mosca, L.; Lynch, V. M.; Zyryanov, G. V.; Anzenbacher, 271 P. Chem. Sci. 2013, 4, 3617. 272

(5) Organisation for the Prohibition of Chemical Weapons, United 273 Nations Joint Mission for the Elimination of the Chemical Weapons 274 Programme of Syrian Arab Republic, http://opcw.unmissions.org/. 275 (6) Review: (a) Royo, S.; Martínez-Máñez, R.; Sancenón, F.; 276 Costero, A. M.; Parra, M.; Gil, S. Chem. Commun. 2007, 4839. E.g.: 277 (b) Gotor, R.; Costero, A. M.; Gil, S.; Parra, M.; Martínez-Máñez, R.; 278 Sancenón, F. Chem.-Eur. J. 2011, 17, 11994. (c) Royo, S.; Costero, A. 279 M.; Parra, M.; Gil, S.; Martínez-Máñez, R.; Sancenón, F. Chem.-Eur. 280 I. 2011, 17, 6931. (d) Nourmohammadian, F.; Wu, T.; Branda, N. R. 281 Chem. Commun. 2011, 47, 10954. (e) Gotor, R.; Royo, S.; Costero, A. 282 M.; Parra, M.; Gil, S.; Martínez-Máñez, R.; Sancenón, F. Tetrahedron 283 2012, 68, 8612. (f) Chulvi, K.; Gaviña, P.; Costero, A. M.; Gil, S.; 284 Parra, M.; Gotor, R.; Royo, S.; Martínez-Máñez, R.; Sancenón, F.; 285 Vivancos, J.-L. Chem. Commun. 2012, 48, 10105. (g) Barba-Bon, A.; 286 Costero, A. M.; Parra, M.; Gil, S.; Martínez-Máñez, R.; Sancenón, F.; 287 Gale, P. A.; Hiscock, J. R. Chem.—Eur. J. 2013, 19, 1586. (h) Walton, 288 I.; Davis, M.; Munro, L.; Catalano, V. J.; Cragg, P. J.; Huggins, M. T.; 289 Wallace, K. J. Org. Lett. 2012, 14, 2686. 290

(7) Review: (a) Burnworth, M.; Rowan, S. J.; Weder, C. Chem.—Eur. 291
J. 2007, 13, 7828. E.g.: (b) Costero, A. M.; Parra, M.; Gil, S.; Gotor, 292
R; Mancini, P. M. E.; Martínez-Máñez, R.; Sancenón, F.; Royo, S. 293
Chem. Asian J. 2010, 5, 1573. (c) Han, S.; Xue, Z.; Wang, Z.; Wen, T. 294
B. Chem. Commun. 2010, 46, 8413. (d) Wu, X.; Wu, Z.; Han, S. Chem. 295
Commun. 2011, 47, 11468. (e) Wild, A.; Winter, A.; Hager, M. D.; 296
Schubert, U. S. Chem. Commun. 2012, 48, 964. (f) Ajami, D.; Rebek, J., 297
Jr. Org. Biomol. Chem. 2013, 11, 3936. 298

(8) Reviews: (a) Diehl, K. L.; Anslyn, E. V. Chem. Soc. Rev. 2013, 42, 299
8596. (b) Askim, J. R.; Mahmoudi, M.; Suslick, K. S. Chem. Soc. Rev. 300
2013, 42, 8649. 301

(9) (a) Rathfon, J. M.; Al-Badri, Z. M.; Shunmugam, R.; Berry, S. M.; 302 Pabba, S.; Keynton, R. S.; Cohn, R. W.; Tew, G. N. *Adv. Funct. Mater.* 303 **2009**, *19*, 689. (b) Climent, E.; Martí, A.; Royo, S.; Martínez-Máñez, 304 R.; Marcos, M. D.; Sancenón, F.; Soto, J.; Costero, A. M.; Gil, S.; 305 Parra, M. *Angew. Chem., Int. Ed.* **2010**, *49*, 5945. (c) Candel, I.; 306 Bernardos, A.; Climent, E.; Marcos, M. D.; Martínez-Máñez, R.; 307 Sancenón, F.; Soto, J.; Costero, A.; Gil, S.; Parra, M. *Chem. Commun.* 308 **2011**, *47*, 8313. (d) Coronado-Puchau, M.; Saa, L.; Grzelczak, M.; 309 Pavlov, V.; Liz-Marzán, L. M. *Nano Today* **2013**, *8*, 461. 310

(10) Real CWA sensing: (a) Sambrook, M. R.; Hiscock, J. R.; Cook, 311
A.; Green, A. C.; Holden, I.; Vincent, J. C.; Gale, P. A. Chem. Commun. 312
2012, 48, 5605. (b) Ordronneau, L.; Carella, A.; Pohanka, M.; 313
Simonato, J.-P. Chem. Commun. 2013, 49, 8946. (c) Pardasani, D.; 314
Tak, V.; Purohit, A. K.; Dubey, D. K. Analyst 2012, 137, 5648. 315
(11) (a) Gómez, T.; Moreno, D.; Díaz de Greñu, B.; Fernández, A. 316

(1) (a) Gominz, T.; Noreno, D.; Diaz de Greñu, D.; Penandez, R.; Sio
 C.; Rodríguez, T.; Rojo, J.; Cuevas, J. V.; Torroba, T. Chem. Asian J. 317
 2013, 8, 1271. (b) Moreno, D.; Díaz de Greñu, B.; García, B.; Ibeas, S.; 318
 Torroba, T. Chem. Commun. 2012, 48, 2994. (c) Moreno, D.; Cuevas, 319
 J. V.; García-Herbosa, G.; Torroba, T. Chem. Commun. 2011, 47, 3183. 320
 (12) Ballesteros, E.; Moreno, D.; Gómez, T.; Rodríguez, T.; Rojo, J.; 321
 García-Valverde, M.; Torroba, T. Org. Lett. 2009, 11, 1269. 322

(13) Phosgene sensors: (a) Kundu, P.; Hwang, K. C. Anal. Chem. 323
2012, 84, 4594. (b) Wu, X.; Wu, Z.; Yang, Y.; Han, S. Chem. Commun. 324
2012, 48, 1895. (c) Feng, D.; Zhang, Y.; Shi, W.; Li, X.; Ma, H. Chem. 325
Commun. 2010, 46, 9203. (d) Ziessel, R.; Ulrich, G.; Harriman, A; 326
Alamiry, M. A. H.; Stewart, B.; Retailleau, P. Chem.—Eur. J. 2009, 15, 327
1359.

(14) Reviews: (a) Anzenbacher, P., Jr.; Lubal, P.; Buček, P.; Palacios, 329
M. A.; Kozelkova, M. E. *Chem. Soc. Rev.* 2010, 39, 3954. (b) Stewart, 330
S.; Ivy, M. A.; Anslyn, E. V. *Chem. Soc. Rev.* 2013, 43, 70. 331
(15) *SIMCA*, version 13, Multivariate Analysis Software; Umetrics 332
AB: Umeå, Sweden 333