1	Synthesis and characterization of a stable humic-urease
2	complex: application to barley seed encapsulation for
3	improving N uptake
4	
5	Running title: Synthesis of humic-urease complex for
6	application to seed encapsulation
7	Beaufray G. Mvila ^a , María C. Pilar-Izquierdo ^a *, María D. Busto ^a , Manuel Perez-
8	Mateos ^a , Natividad Ortega ^a
9	^a Department of Biotechnology and Food Science, Area of Biochemistry and Molecular
10	Biology, University of Burgos, Plaza Misael Bañuelos, s/n, E-09001 Burgos, Spain
11	* Corresponding author. Phone number: (0034) 947258800. Fax number: (0034)
12	947258831. E-mail address: mcpilar@ubu.es
13	
14	

15 Abstract

BACKGROUND: Most of the N fertilizers added to the soil are not efficiently used by plants and are lost to the atmosphere or leached from the soil, causing environmental pollution and increasing cost. The barley seed encapsulation in calcium alginate gels containing free or immobilized urease was investigated to enhance plant utilization of soil N.

21 RESULTS: Urease was immobilized with soil humic acids (HA). A central composite 22 face-centered design was applied to optimize the immobilization process, reaching an 23 immobilization yield of 127%. Soil stability of urease was enhanced after the 24 immobilization. The seed encapsulation with free urease (FU) and humic-urease 25 complex (HUC) resulted in an urease activity retention in the coating layer of 46 and 26 24%, and in germination rates of 87 and 92%, respectively. Under pot culture 27 conditions, the pots planted with seeds encapsulated with FU and HUC showed higher 28 ammonium N (NH $_{4}^{+}$ -N) (26 and 64%, respectively) than the control soil at 28 days after 29 planting (DAP). Moreover, the seed encapsulation with FU and HUC increased the N 30 uptake 83 and 97%, respectively, at 35 DAP.

31 CONCLUSION: The seed encapsulation with urease could substantially contribute to32 enhance the plant N nutrition in the early stages of seedling establishment.

33 KEYWORDS: Enzyme immobilization; Nitrogen availability; Plant growth; Seed
 34 coating; Seed encapsulation; Soil urease

36 INTRODUCTION

37 The application of N fertilizer is a common practice to increase crop productivity, 38 since N is a limiting factor for plant growth. The global consumption of N fertilizer is 39 closely related with the cereal production. Thus, in example, of the total N fertilizers 40 consumed in world in 2010/11, 55.2% it is estimated to have been applied to cereal 41 production.¹ Barley ranks fourth among the cereals in worldwide production. In Spain is 42 one of the most important crops, representing up to 40% of the total cereal production.² 43 However, most of N fertilizers added to the soil are not efficiently utilized by plants. 44 For example, the N use efficiency (NUE) of N fertilizers in the global cereal production 45 had been estimated at 33%.³ Moreover, the N fertilizers that are not used by plants are 46 lost to the atmosphere or leached from the soil which, besides increasing production 47 cost, has adverse effects on the environment including greenhouse effect and surface and ground water pollution.⁴ 48

49 Nitrogen occurs in soils in both inorganic and organic forms. Although there is 50 evidence that soluble organic forms of N (such as amino acids) can be taken up by 51 plants, inorganic N forms (NO₃⁻ and NH₄⁺) have classically been considered to dominate plant uptake.⁵ In this way, amidohydrolases play an important role in organic 52 53 N mineralization in soils and thus, N supply for plants. These enzymes catalyze the hydrolysis of C-N bond other that peptide bonds in linear amides.⁶ Among 54 55 amidohydrolases, urease is one of the most studied enzymes. Urease is a ubiquitous 56 enzyme in the soil, where it is produced by a vast number of both eukaryotic and 57 prokaryotic organisms. This enzyme catalyzes the hydrolysis of urea to CO₂ and NH₃. 58 Urea is commonly found in natural environment derived from urine excretion by 59 animals (animal wastes) and the decomposition of nitrogenous compounds (i.e. amino

acids, purines) from dead organisms.⁷ Moreover, urea is quantitatively the most
important N fertilizer.⁵ In this way, urease activity of soil has a great importance in
agriculture due to urea serves as a readily available N source for the plant growth.

63 Enhance the NUE by plants is essential for a sustainable agriculture. Thereby, an 64 increase in the rate of utilization of soil organic N in the plant rhizosphere could 65 improve the NUE, avoiding or reducing the application of N fertilizers. This increase 66 may be achieved by means of seed coating with stabilized enzymes. The seed coating 67 allows the application of "materials" onto the natural seed coat in such a way that they can affect the seed or soil at the seed-soil interface.⁸ Enzymes can be stabilized by their 68 69 association with humic colloids, which protect them against proteolytic activity and other processes leading to its inactivation in soil.^{9,10} Pilar-Izquierdo et al.^{11,12} reported 70 71 that the seed coating with alkaline phosphatase stabilized by immobilization in soil 72 humates enhanced the utilization by plants of accumulated soil P. A similar strategy, 73 using stabilized urease, may be helpful to mineralize naturally occurring soil organic N, by hydrolysis of urea compounds to ammonia, and benefit N uptake by plants. Among 74 75 the techniques that can be used for seed coating (film-coating, pelleting, encrusting and encapsulation), 8,13,14 the encapsulation in calcium alginate capsules¹² have been 76 successfully applied for the coating of barley seeds.^{12,15} By means of encapsulation the 77 78 coating material is enclosed inside the calcium alginate capsule formed around the seed and, after seed drying, a uniform film adhered to the seed surface is formed.¹² Applying 79 this technique no coating material losses or seed agglomeration have been observed,¹⁵ 80 81 two important problems that can occur when seeds are coated.

82 In this context, the aim of this work was to synthesize a stable HUC and to study the 83 effect of the encapsulation of barley seeds in calcium alginate gels containing free or 84 immobilized urease on N uptake and plant growth. Specifically, this research can be 85 very valuable in order to develop "fertilizers" enriched with stabilized enzymes as a 86 means of bringing about–localized solubilization of soil organic N, avoiding the 87 deleterious effects of excessive application of N chemical fertilizers.

88 MATERIALS AND METHODS

89 Experimental design

90 Urease (urea amidohydrolase EC 3.5.1.5) from jack bean (Canavalia ensiformis) 91 (Calbiochem, San Diego, CA) was immobilized by co-flocculation with HA and Ca^{2+} . 92 A central composite face-centered star points design (CCF) was applied to evaluate the 93 effect of immobilization parameters. The properties and stability of the HUC were 94 studied. Barley seeds (Hordeum vulgare L., var. vanessa) were coated by encapsulation 95 in Ca-alginate gels containing FU or HUC. A germination test was performed to assess 96 the viability of the encapsulated seeds. A pot culture experiment was carried out to 97 study the effect of the coating on the plant growth and N uptake. The soil used in the 98 experiment was collected from the farm field surface of 0-10 cm at Ribera del Arlanza 99 in Burgos (Spain). Before use, the soil was air-dried at room temperature and then 100 gently ground to pass through a 2-mm sieve. Relevant soil properties are: pH 7.0, total C 26.8 g kg⁻¹, organic C 20.7 g kg⁻¹ and total N 13.2 g kg⁻¹. 101

102 Non-treated and treated seeds were germinated on a wet filter paper at 26 °C in a 103 growth chamber (Conviron[®] E7) for 96 h. For each type of seeds, five seedlings were 104 transplanted to a plastic pot (containing 150 g of non-amended soil) at 10 mm below the 105 soil surface, and then the pots were transferred to a controlled environment chamber. 106 The plants were grown under environmental conditions of 20/15°C photo/dark period 107 temperature and 16 h photoperiods per day. The pots were watered everyday with 15 ml 108 of deionized water. Each experiment was conducted five times. Five plants (one plant of 109 each replicate) were carefully freed from soil at 7, 14, 21, 28 and 35 DAP. After each 110 harvest the plant growth was determined by measuring length, dry weight and N 111 concentration of the shoot. Parallel, soil samples were taken in order to evaluate NH4⁺-112 N and urease activity in the bulk soil. The inorganic N of soil was extracted according to Keeny and Nelson¹⁶ and the NH₄⁺-N present in the solution was evaluated following 113 the method proposed by Keeney and Bremner¹⁷ and modified by Beck.¹⁸ 114

115 Enzyme immobilization

Urease was immobilized by co-flocculation with HA and Ca²⁺. The HA were 116 obtained according to Busto et al.¹⁹ The HUC were obtained as described by Pilar et al. 117 ²⁰ with some modifications. Briefly, 2 mL of HA (ranging from 2 to 6 mg mL⁻¹) was 118 added to 2 mL of the urease solution (0.25 mg mL⁻¹) and 3 mL of phosphate-citrate-119 120 borate buffer (PCBB: 100 mM phosphate, 57 mM borate, 36 mM citrate) at different pH 121 value (6.0-10.0). The resulting immobilization solution was mixed by orbital shaking 122 (150 rpm) at different agitation times (from 30 to 180 min) and temperatures (from 4 to 123 50 °C). The mixture was flocculated by the addition of 2 mL of 0.5 M CaCl₂ and shaken 124 gently (150 rpm) for 6 h at 20 °C. The suspension was centrifuged (15400 g) for 15 min 125 at 4 °C and the pellet was washed five times with 0.1 M calcium acetate buffer (pH 4.5). 126 The insoluble complexes were resuspended in 10 mL PCBB (pH 7) and assayed for 127 urease activity. The immobilization yield (IY, %) was determined as follows:

128 IY (%) =
$$(A_{HUC} - A_H) \times 100/Ao$$
 (1)

where A_{HUC}, A_H and A₀ were the urease activity of the HUC, the HA and the activity
added for immobilization, respectively.

131 Assay of urease activity

132 Urease activity was measured following the method reported by Kandeler and 133 Gerber²¹ with some modifications. Briefly, the activity was determined by incubating 1 134 mL of 50 mM urea (Sigma, St. Louis, MO), 3.5 mL of PCBB (pH 7) and 0.5 mL of the 135 enzyme solution at 37 °C for 15 min. For the urease activity of coated seeds and soil, 136 samples of five seeds and 1 g of fresh soil, respectively, were added to 4 mL of PCBB 137 (pH 7). After the reaction, 15 mL of 1 M KCl and 10 mM HCl was added to the mixture 138 and shaken gently (150 rpm) for 30 min at room temperature to extract the NH4⁺-N 139 released from urea. Then, 2 mL of 0.5 M CaCl₂ was added (to flocculate the organic 140 matter) to the mixture. The resulting solution was shaken gently (150 rpm) for 30 min at 141 room temperature and filtered (Whatman No. 6), and the NH₄⁺-N present in the filtrate was determined. In parallel, a standard curve of NH4⁺-N in the range of 0-20 µg mL⁻¹ 142 143 was run. One enzymatic unit was defined as the amount of urease that produces 1 µg of 144 NH₄⁺-N from urea over 1 h under the assay conditions.

145 **Properties and stability of the immobilized urease**

The Michaelis constants (K_m) of FU and HUC were determined by measuring initial rates of the reaction at urea concentrations ranging from 6 to 30 mM. The K_m values were calculated by using a Lineweaver-Burk plot. The effect of pH on FU and HUC activity was studied using PCBB with pH values ranging from 4 to 10. The optimum temperature for hydrolysis of urea was determined by measuring the urease activity at incubation temperatures from 30 to 80° C.

To test the stability in soil, FU (0.0375 mg g^{-1} soil; 1190 U g^{-1} soil) and HUC (0.06 g g^{-1} soil; 1684 U g^{-1} soil) were incorporated in samples of soil described above. The soil was held at 20°C for 30 days and the residual urease activity measured periodically. To calculate the activity of FU and HUC added to soil, activity of endogenous urease was also measured and subtracted. The residual activity of the FU and HUC was calculated as a percentage of the initial activity measured in soil after their addition.

158

Seed coating and germination test

159 Barley seeds were encapsulated in Ca-alginate gels containing FU and HUC using the method described by Pilar et al.¹⁵ (modified from Patel et al.²²) and modified as 160 161 follow: the encapsulation solution contained 6% (w/v) carboxymethylcellulose (CMC) 162 (Sigma, St. Louis, MO), 2% (w/v) calcium chloride and different enzyme/CMC solution 163 ratios (FU: 0.1, 0.2 and 0.4 mg mL⁻¹; HUC: 0.1 g mL⁻¹). The coating urease activity was 164 defined by the ratio of the urease activity of the encapsulated seeds to the activity of the 165 free or immobilized enzyme used in the coating process (FU: 1159 U mg⁻¹; HUC: 213 U g-1). Determinations were replicated five times. The germination test was assayed as 166 described by Pilar-Izquierdo et al.¹¹ 167

168 **Plant analysis**

Length, dry weight and shoot N concentration were determined. Dry weight was determined by oven-drying the shoots at 105 °C for 48 h. The inorganic N content was evaluated following the method of Hevia and Cioccia.²³ Plant N uptake was calculated by multiplying the dry weight by shoot N concentration.

173 Statistical methods

174 The immobilization parameters were optimized using response surface 175 methodology.²⁴ A CCF was employed in this regard. Optimized conditions and response 176 surfaces were calculated and drawn, respectively, with Statgraphics Centurion 16.1.07. 177 This software package was also used to fit the second-order model to the independent178 variables by using eq 2

179
$$y = b_0 + \sum_{i=1}^k b_i X_i + \sum_{i=1}^k b_{ii} X_i^2 + \sum_{i=1}^{i < j} \sum_j b_{ij} X_i X_j + e$$
(2)

180 where y is the dependent variable (response variable) to be modelled, X_i and X_j are the 181 independent variables (factors), b_o , b_i , b_{ii} and b_{ij} are regression coefficients and e is the 182 error. The model was simplified by dropping terms which were not statistically 183 significant (p>0.05) by analysis of variance (ANOVA).

Analysis of variance and regression analysis were performed by Statgraphics Centurion 16.1.07 for Windows. The standard errors of the mean (SEM) are given in Tables and Figures. The standard errors of difference (SED) were calculated to test differences in urease activity of the soil, NH4⁺-N in soil and plant growth between encapsulated seeds and control (soil planted with non-treated seeds).

189 **RESULTS**

190 Urease immobilization

191 Preliminary experiments were carried out to screen the parameters that influence the 192 preparation of the HUC and the experimental domain. From these experiments, four 193 factors were investigated: temperature (X_1) , HA concentration (X_2) , immobilization pH 194 (X_3) and contact time (X_4) . Table 1 presents the experiment matrix, together with the 195 experimental results. The immobilization yield ranged from 53 to 116%. High 196 percentages of immobilization (>110%) were reached when a low level of temperature 197 was used at high or central values of both HA concentrations, immobilization pH and contact time (runs 5 and 9, respectively). 198

Analysis of variance (ANOVA) was used to determine the adequacy and the significance of the quadratic model. The R^2 value of the model was 0.92, indicating that 92% of the variability in the response can be explained by the model. The adjusted R^2 value of 0.82 suggested that the model was significant. A very low value of the coefficient of variation (CV = 4.44%) clearly indicated a very high degree of precision and a good reliability of the experimental values.

A regression analysis was performed in order to determine the coefficients of the significant effects of each response variable. The effects showing less than 95% significance were omitted. In this sense, eq 3 explains the data obtained in this experiment.

209 IY (%) = -79.794 -0.217
$$X_1$$
+ 0.036 (X_1)²-0.169 X_1X_3 - 0.003 X_1X_4 -
210 - 4.796 (X_2)²+0.043 X_2X_4 -0.037 X_3X_4 (3)

211 The response surface plots for the immobilization yield showed an increase as the 212 temperature was decreased and a slight increase when the immobilization time 213 increased (Figure 1a). The effect of the HA concentration and pH on the response at a 214 fixed temperature and time of 4 °C and 180 min is illustrated in Figure 1b. The response value reached its highest level at 4.5 mg mL⁻¹ HA, whereas pH showed a maximum at 215 216 pH 8.0. Under these conditions, the model predicted an immobilization yield of 127%. 217 To confirm this result, a validation assay was conducted in the conditions imposed at 218 the optimum. In this assay an immobilization yield of 121% was obtained. This value is 219 in good agreement with the predicted values for the analyzed response, validating the 220 mathematical models attained in the studied region.

221 Properties and stability of the immobilized enzyme

222 Figure 2 shows Lineweaver-Burk plots for FU and HUC. The FU followed 223 Michaelis-Menten kinetics over the range of substrate concentration studied. However, 224 the HUC did not follow pure Michaelis-Menten kinetics. The shape of this graph was typical of enzymes exhibiting substrate inhibition at high substrate concentration.²⁵ In 225 226 this case, the determination of apparent K_m was possible by using the linear portion of the curve at low substrate concentration.²⁶ The K_m value for the HUC was higher (129.5 227 228 mM) than that of the FU (19.5 mM), increasing it by a factor of 6.6. The pH activity 229 profile of both FU and HUC is plotted in Figure 3a. The FU showed optimal activity at 230 pH 5.5, whereas the optimum pH of the HUC was determined at pH 6.2.

The dependence of the urease activity on temperature is shown in Figure 3b. The optimum temperature of both FU and HUC was found at 60°C. At temperature values of 70 and 80°C, the HUC showed the activities of about 8.4 and 9.4% higher than FU, respectively, which can be due to the protective effect of the immobilization support.

The stability of both FU and HUC in soil was studied (Fig. 4). The FU was almost inactivated after 3 days, whereas the HUC retained about 35% and 13% of its initial activity after 7 and 30 days, respectively, of its addition to soil.

238 Seed encapsulation and germination test

The effect of the barley seed encapsulation with different FU concentration on the coating urease activity was studied. When encapsulation was performed with enzyme/CMC solution ratios of 0.1 and 0.2 mg mL⁻¹ the urease activity retained in the coating layer was similar (31 and 33%, respectively). However, at the highest enzyme concentration assayed (0.4 mg mL⁻¹), a significant increase (p < 0.01) in the coating urease activity was observed (46%). The seed encapsulation with immobilized urease was carried out at a fixed HUC/CMC solution ratio of 0.1 g mL⁻¹. Under this assay condition, the coating urease activity was 24%. This enzyme concentration was chosen in order to achieve an adequate level of urease activity in the coating layer. In this way, Pilar-Izquierdo *et al.*¹² found that the seed encapsulation with higher concentrations of humic-enzyme complexes resulted in a remarkable increase in the thickness of the seedcoating layer, which limited the substrate diffusion. As far as the germination of seeds is concerned, the seeds encapsulated with FU and HUC germinated 88% (for all enzyme concentration assayed) and 92%, respectively, in comparison with non-treated seeds.

253 Effect of seed encapsulation on urease activity and NH4⁺-N of soil

254 The effect of seed encapsulation on urease activity and NH₄⁺-N of the soil is shown 255 in Table 2. A significant increase (p < 0.05) in the urease activity in the bulk soil of pots 256 planted with encapsulated seeds was observed at the end of the study period. Thus, at 35 257 DAP, the soil urease activity of pots planted with seeds encapsulated with FU and HUC 258 was increased 15 and 21%, respectively, in comparison with control soil. No significant 259 differences between the seed coating with FU and HUC were found. The NH4⁺-N in soil 260 was increased with time to maximum rates at 28 DAP in all pots, and then decreased to 261 a concentration similar to that at the start of the experiment. Moreover, this increase was 262 significantly higher in pots planted with treated seeds from 21 DAP. Thus, the NH4⁺-N 263 in soil planted with seeds encapsulated with FU and HUC increased 56 and 21%, 264 respectively, at 21 DAP, and 26 and 64%, respectively, at 28 DAP. However, at the end 265 of the study period, only the pots planted with seeds encapsulated with HUC showed a 266 significant increase (23%) in the NH_4^+ -N in soil.

267 Effect of seed encapsulation on plant growth and N uptake

268 The effect of seed encapsulation with FU and HUC on plant growth and N uptake is 269 shown in Table 3. The seed encapsulation significantly increased the shoot length 270 during the first four weeks of plant growth. Thus, at 7 DAP, plants grown from seeds 271 encapsulated with both FU and HUC showed higher shoot length (38%) than the 272 control. However, although control plants grew more slowly, they reached a shoot 273 length similar to the plants grown from encapsulated seeds at the end of the study 274 period. The shoot dry weight of plants grown from coated seeds, like the shoot length, 275 was significantly increased (p < 0.01) during the first three weeks of growing. Moreover, 276 the seed encapsulation with FU increased the shoot dry weight 15%, in comparison 277 with control plants, at the end of the study period. The shoot N content was significantly 278 higher (p = 0.00) in plants grown from seeds encapsulated with FU and HUC than in 279 control plants, showing an increase of 55 and 87%, respectively, after 35 DAP. 280 Moreover, this increase was significant higher in plants grown from seeds encapsulated 281 with HUC than that with FU since 21 DAP to the end of the experiment. A decrease in 282 the shoot N concentration with time was observed in all pots, which as expected, since is known that the N content of tissues decreases with the life cycle of plant.²⁷ Finally, 283 284 the seed encapsulation with urease significantly increased (p = 0.000) the N uptake by 285 plants. Thus, at 35 DAP, the plants grown from seeds encapsulated with FU and HUC 286 showed an increase in the N uptake of 83 and 97%, respectively.

287 **DISCUSSION**

Stabilized extracellular soil enzymes, in particular hydrolases such as urease, play an important role in the availability of nutrients for plants. Soil enzymes can be stabilized through association with clay minerals, particulate organic matter and HA.^{10,30} In this way, soil urease is crucial for the N cycle and its association with HA is a fundamental
 requirement for its stability.²⁸

293 The chemical complexity of the humic substances still prevents a full understanding 294 of their interactions with reactive biomolecules presents in the environment, such as extracellular enzymes.²⁹ However, several mechanisms have been proposed for the 295 296 stability of humic-enzyme complexes. Thus, humic compound may adsorb enzymes by ion exchange, covalent complexation and hydrogen bonding.³⁰ Moreover, enzymes can 297 298 be trapped within the humic net and also immobilized at the surface by adsorption 299 forces.³¹ Recent studies applying nuclear magnetic resonance spectroscopy indicate that enzymes are stabilized by encapsulation in humic aggregates³² or by non-covalent 300 interactions, such as van der Waals, H-bonds and π - π bonds.²⁹ Anyway, insufficient 301 302 data are available to draw definitive conclusions about the nature of the interaction in 303 the samples examined here.

304 It is well known that immobilization can affect enzyme activity and catalytic 305 properties. In the present study, the binding of urease to soil HA increased the enzyme 306 activity, reaching immobilization yields higher than 100% under optimal conditions. Busto et al.¹⁹ also observed immobilization yields higher than 100% in humate-β-307 glucosidase and humate-carboxymethyl cellulase complexes by 180 and 350%, 308 respectively. In contrast, Pilar et al.¹⁸ reported a loss of activity in phosphatase-humate 309 310 complexes with respect to the initial enzyme concentration. The enzymes attached to the 311 humic matrix have a microenvironment quite different than that of free enzymes in solution,²³ because they are affected by different diffusion, charge and steric effects.³¹ 312 313 Therefore, the activity of the humic-enzyme complexes toward substrates may be 314 different from that of the free enzyme.

315 With regard to the catalytic properties of the immobilized urease, the increase in the K_m value observed after immobilization indicate a lower affinity of the immobilized 316 317 enzyme by its substrate in comparison with the soluble enzyme, which may be caused 318 by steric hindrance of the active site by the support, the loss of enzyme flexibility 319 necessary for substrate binding or by diffusion limitations of the substrate.³³ In general, enzymes bound to soil components have higher K_m values than the free enzymes.³⁴ The 320 321 shift in the pH-activity curves towards alkaline region observed upon immobilization has been attributed to the microenvironment created by the charged carriers. Tan³⁵ 322 323 reported that dissociation of COOH in humic matter starts at pH 3.0, and the humic 324 molecule becomes negatively charged. As a result of a negative surface charge the pH 325 in the microenvironment of the enzyme will be lower than the bulk pH. A higher bulk 326 pH is required in providing an optimum pH in the microenvironment of the enzyme and hence a shift to higher values is encountered.³⁶ The higher stability of the immobilized 327 328 urease added to the soil, in comparison with their soluble counterpart, suggests that the 329 association of the enzyme with the HA protects it from degradation in soil. Therefore, 330 given the increase in the activity and stability of urease immobilized in the humic 331 support, the seed coating with the HUC synthesized is expected to improve the NUE in 332 the plant rhizosphere.

The values of enzyme activity retained to the seed coating layer may be related with the coating process. Firstly, a leakage of the enzyme can occur during the encapsulation process, probably before the alginate capsule was formed,³⁷ which could result in a decrease in the calculated coating yield. Thus, Vikartovska *et al.*³⁸ reported enzyme losses of about 56% during the first step of the encapsulation of glucose oxidase in polyelectrolyte complex capsules. Moreover, the enzyme added during the seed encapsulation is enclosed inside the capsule formed, which could cause diffusional
 restrictions so that not all the activity of the immobilized enzyme was expressed.³⁹

341 Germination of coated seeds can be affected by the coating composition, coat 342 thickness and species, among others, which may increase, decrease or not affect the 343 germination rates.⁴⁰ In this work, a slight decrease in the germination rate of 344 encapsulated seeds was observed. These results were similar to those reported by Pilar-Izquierdo et al.¹² when barley seeds were encapsulated with phosphatase free and 345 346 immobilized in HA. This decrease in the germination rate could be attributed to the 347 coating acting as an oxygen barrier or the hardness of the Ca-alginate capsule formed 348 around the seed. Film-coating polymers decreased seedling emergence in wet soils with low oxygen diffusion rate.⁴¹ Moreover, a delay of 24 h in germination of encapsulated 349 seeds in comparison with control was observed. In this way, Ester et al.⁴² found that the 350 351 germination of seed film-coated with an insecticide was delayed, although after a few 352 days the growth of plants from film-coated seeds was similar to control plants. Since the 353 FU/CMC solution ratio had no effect on the germination percentage of encapsulated seeds, 0.4 mg mL⁻¹ was selected as the optimal enzyme concentration taking into 354 355 account the coating urease activity.

Increases in NH₄⁺-N concentration with time in soils incubated under laboratory conditions have been reported.^{4,43} Moreover, Abbasi *et al.*⁴ also observed a similar trend in the NH₄⁺-N concentration in a control soil, in which this concentration increased with time to a maximum and then declined to background level at the end of the experiment. The increase in the NH₄⁺-N with time indicates that the mineralization of organic N present in soil has occurred. Moreover, the significantly higher NH₄⁺-N concentration observed in pots planted with encapsulated seeds from 21 DAP, in comparison with 363 control soils, may be attributable to the urease activity added in the seed coating layer. 364 Thus, although increases in the urease activity of bulk soil were only detected at the end 365 of the study period, a higher rhizosphere urease activity in the first weeks of the pot 366 experiment could be expected. On the other hand, the decrease in the soil NH4⁺-N 367 concentration observed in all pots at the end of the study period could be partially due to 368 the nitrification of NH₄⁺-N to NO₃⁻-N. Nitrification is often considered the major fate of NH_4^+ in agricultural soil.⁴⁴ Thus, increases in the NO₃-N in soils incubated under 369 laboratory conditions have been reported.^{4,45} Furthermore, the decrease in the soil NH₄⁺-370 371 N could be also due to the increase observed in the plant N uptake. In addition, the changes in the NH₄⁺-N concentration observed in the pots with time could be related 372 with the microbial activity.44,46 373

374 CONCLUSION

375 In this research the synthesis of a stable HUC and the application of FU or HUC to 376 barley seed encapsulation for improving plant utilization of soil N was studied. The 377 results obtained in the pot experiment showed the beneficial effect of the seed 378 encapsulation by increasing the shoot N content and the N uptake. No significant 379 differences in the N uptake of plants grown from seeds encapsulated with FU and HUC 380 were found at the end of the study period. Nevertheless, althought the shoot N content 381 in plants grown from seeds encapsulated with HUC was higher than in plants grown 382 from seeds encapsulated with FU, at 35 DAP, the value of N content was in the same 383 order of magnitude. These results, together with the higher NH₄⁺-N concentration found 384 in the pots planted with encapsulated seeds, could be explained by a higher urease 385 activity in the root vicinity as a consequence of the coating applied on the seeds.

The seed encapsulation with urease could potentially enhance the N plant nutrition, without the massive addition of N fertilizers. However, further research to evaluate the NUE is necessary. Moreover, it would be necessary to carry out field experiments to evaluate the effect of the seed encapsulation with urease on the plant growth and grain yield under natural conditions.

391 ACKNOWLEDGEMENTS

392 We thank to the "Ministere de L'Enseignement Superieur de la Republique du Congo"

393 by the fellowship to BG Mvila.

394 REFERENCES

- Heffer P, Assessment of fertilizer use by crop at the global level 2010-2010/2011.
 International Fertilizer Industry Association (IFA). Paris, France. (2013).
- Mateo EM, Gimeno-Adelantado JV, Soria JM, García-Esparza MA, Mateo-Castro
 M and Jiménez M, Barley, a potential source of ochratoxin A in food in the
 framework of climate change, in *Industrial, Medical and Environmental Applications of Microorganisms*, ed. by Méndez-Vilas A. Wageningen Academic
- 401 Publishers, The Netherlands, pp. 255-260 (2014).
- 402 3 Raun WR and Johnson GV, Improving nitrogen use efficiency for cereal
 403 production. *Agron J* 91:357-363 (1999).
- 404 4 Abbasi MK, Hina M and Tahir MM, Effect of *Azadirachta indica* (neem), sodium
 405 thiosulphate and calcium chloride on changes in nitrogen transformations and
 406 inhibition of nitrification in soil incubated under laboratory conditions.
 407 *Chemosphere* 82:1629-1635 (2011).

- 408 5 Miller AJ and Cramer MD, Root nitrogen acquisition and assimilation. *Plant Soil*409 274:1-36 (2004).
- 410 6 Ekenler M and Tabatabai MA, Arylamidase and amidohydrolases in soils as
 411 affected by liming and tillage systems. *Soil Till Res* 77:157-168 (2004).
- Wang WH, Köhler B, Cao FQ and Liu LH, Molecular and physiological aspects
 of urea transport in higher plants. *Plant Sci* 175:467-477 (2008).
- 414 8 Scott JM, Seed coatings and treatments and their effects on plant establishment.
 415 Adv Agron 43:43-83 (1989).
- Burns RG, Interaction of enzymes with soil mineral and organic colloids, in *Interactions of Soil Minerals with Natural Organics and Microbes*, ed. by Huang
 PM and Schnitzer M. Soil Science Society of America, Madison, pp. 429-451
 (1986).
- Burns RG, Deforest JL, Marxsen J, Sinsabaugh RL, Stromberger ME, Wallenstein
 MD, Weintraub MN and Zoppini A, Soil enzymes in a changing environment:
 Current knowledge and future directions. *Soil Biol Biochem* 58:216-234 (2013).
- Pilar-Izquierdo MC, Ortega N, Perez-Mateos M and Busto MD, Barley seed
 coating with free and immobilized alkaline phosphatase to improve P uptake and
 plant growth. *J Agr Sci* 150:691-701 (2012).
- 426 12 Pilar-Izquierdo MC, Busto MD, Ortega N and Perez-Mateos M, Seeds
 427 encapsulated in calcium-alginate gels with phosphatase and humate-phosphatase
 428 complexes for improving phosphorus bioavailability. *Agron J* 105:1565-1570
 429 (2013).

- Taylor AG, Seed treatments, in *Encyclopedia of Applied Plant Sciences*, ed. by
 Thomas B, Murphy DJ and Murray BG. Elsevier Academic Press, Amsterdam,
 pp. 1291-1298 (2003).
- 433 14 Sarrocco S, Raeta R and Vannacci G, Seeds encapsulation in calcium alginate
 434 pellets. *Seed Sci Technol* 2:649–661 (2004).
- 435 15 Pilar MC, Ortega N, Perez-Mateos M and Busto MD, Alkaline phosphatase436 polyresorcinol complex: characterization and application to seed coating. *J Agr*437 *Food Chem* 57:1967-1974 (2009).
- 438 16 Keeney DR and Nelson DW, Nitrogen-inorganic forms, in *Methods of Soil*439 *Analysis. Part 2 Chemical and Microbiological Properties (2nd Edition)*, ed. by
- 440 Page AL, Miller RH and Keeney DR. American Society of Agronomy, Madison,
 441 pp. 643-698 (1982).
- Keeney DR and Bremner JM, Comparison and evaluation of laboratory methods
 of obtaining an index of soil nitrogen availability. *Agron J* 58:498-503 (1966).
- 444 18 Beck TH, Die N-Mineralisierung von Böden im Laborbrutversuch. Z Pflanz
 445 Bodenkunde 146:243-252 (1983).
- Busto MD, Ortega N and Perez-Mateos M, Stabilisation of cellulases by
 crosslinking with glutaraldehyde and soil humates. *Bioresource Technol* 60:27-33
 (1997).
- Pilar MC, Ortega N, Perez-Mateos M and Busto MD, Kinetic behaviour and
 stability of *Escherichia coli* ATCC27257 alkaline phosphatase immobilised in soil
 humates. *J Sci Food Agr* 83:232-239 (2003).
- 452 21 Kandeler E and Gerber H, Short-term assay of soil urease activity using
 453 colorimetric determination of ammonium. *Biol Fert Soils* 6:68-72 (1988).

454	22	Patel AV, Pusch I, Mix-Wagner G and Vorlop KD, A novel encapsulation
455		technique for the production of artificial seeds. Plant Cell Rep 19:868-874
456		(2000).

- 457 23 Hevia P and Cioccia AM, Application of a colorimetric method of the
 458 determination of nitrogen in nutritional studies with rats and humans. *Nutr Rep Int*459 38:1129-1136 (1988).
- 460 24 Ortega N, Perez-Mateos M, Pilar MC and Busto MD, Neutrase immobilization on
 461 alginate-glutaraldehyde beads by covalent attachment. *J Agr Food Chem* 57:109462 115 (2009).
- 463 25 Busto MD, Ortega N and Perez-Mateos M, Studies on microbial β-D-glucosidase
 464 immobilised in alginate gel beads. *Process Biochem* 30:421-426 (1995).
- 465 26 Parr SR, Some kinetic properties of the β -D-glucosidase (cellobiase) in a 466 commercial cellulase product from *Penicillium funiculosum* relevance in the 467 hydrolysis of cellulose. *Enzyme Microb Tech* **5**:448-462 (1983).
- 468 27 Barbazán M, Ferrando M and Zamalvide JP, Dry matter accumulation and
 469 nitrogen in anual Winter grass used as cover crops in vineyards. *Agrociencia*470 6:10-19 (2002) (In Spanish).
- 471 28 Dong LH, Yang JS, Yuan HL, Wang ET and Chen WX, Chemical characteristics
 472 and influences of two fractions of Chinese lignite humic acids on urease. *Eur J*473 *Soil Biol* 44:166-171 (2008).
- 474 29 Mazzei P, Oschkinat H and Piccolo A, Reduced activity of alkaline phosphatase
 475 due to host-guest interactions with humic superstructures. *Chemosphere* 93:1972476 1979 (2013).

477	30	Nannipieri P, Sequi P and Fusi P, Humus and enzyme activity, in Humic
478		Substances in Terrestrial Ecosystems, ed. by Piccolo A. Elsevier, Amsterdam, pp
479		293-328 (1996).

- 480 31 Ruggiero P, Dec J and Bollag JM, Soil as a catalytic system, in *Soil Biochemistry*,
 481 *vol. 9*, ed. by Stotzky G and Bollag JM. Marcel Dekker, New York, pp. 79-122
 482 (1996).
- 483 32 Tomaszewski JE, Schwarzenbach RP and Sander M, Protein encapsulation by
 484 humic substances. *Environ Sci Technol* 45:6003-6010 (2011).
- 485 33 Yan J, Pan G, Ding C and Quan G, Kinetic and thermodynamic parameters of β-
- 486 glucosidase immobilized on various colloidal particles from a paddy soil. *Colloids*487 *Surface B* 79:298-303 (2010).
- 488 34 Kaleeswari RK, Role of phosphatase enzymes in phosphorus nutrition of crops.
 489 Agric Rev 28:149-153 (2007).
- 490 35 Tan KH, Chemical processes, in *Handbook of Processes and Modeling in the*491 *Soil-Plant System*, ed. by Benbi DK and Nieder R. Haworth Press, Binghamton,
 492 pp. 27-56 (2003).
- 493 36 Reshmi R, Sanjay G and Sugunan S, Immobilization of α-amylase on zirconia: A
 494 heterogeneous biocatalyst for starch hydrolysis. *Catal Commun* 8:393-399 (2007).
- 495 37 Kurayama F, Suzuki S, Bahadur NM, Furusawa T, Ota H, Sato M and Suzuki N,
- 496 Preparation of aminosilane-alginate hybrid microcapsules and their use for
 497 enzyme encapsulation. *J Mater Chem* 22:15405-15411 (2012).
- 498 38 Vikartovská A, Bučko M, Mislovičová D, Pätoprstý V, Lacík I and Gemeiner P,
 499 Improvement of the stability of glucose oxidase via encapsulation in sodium

- alginate-cellulose sulphate-poly (methylene-co-guanidine) capsules. *Enzyme Microb Tech* 41:748–755 (2007).
- 502 39 Tanriseven A and Doğan S, Immobilization of invertase within calcium alginate
 503 gel capsules. *Process Biochem* 36:1081–1083 (2001).
- Gorim L and Asch F, Effects of composition and share of seed coatings on the
 mobilization efficiency of cereal seeds during germination. *J Agron Crop Sci* **198**:81-91 (2012).
- 507 41 Kim SH and Taylor AG, Germinability of film-coated snap bean seed as affected
 508 by oxygen diffusion rate under different soil moisture contents. *Korean J Crop Sci*509 49:46–51 (2004).
- 510 42 Ester A, de Vogel R and Bouma E, Controlling *Thrips tabaci* (Lind.) in leek by
 511 film-coating seeds with insecticides. *Crop Prot* 16:673-677 (1997).
- 43 Yanyu S, Changchun S, Guisheng Y, Yingchen L, Rong M and Jiaoyue W,
 513 Effects of N addition on soil enzyme activities in marshland ecosystem of
 514 Northeast China: an incubation experiment. *International Conference on*515 Agricultural and Biosystems Engineering. Advances in Biomedical Engineering 1516 2:5-8 (2011).
- 517 44 Robertson GP and Groffman PM, Nitrogen transformations, in *Soil Microbiology*,
 518 *Ecology and Biochemistry (4th Edition)*, ed. by Paul EA. Elsevier Academic Press,
 519 Amsterdam, pp. 421-446 (2015).
- Abbasi MK, Shah Z and Adams WA, Mineralization and nitrification potentials of
 grassland soils at shallow depth during laboratory incubation. *J Plant Nutr Soil Sc*164:497-502 (2001).

523 46 Myrold DD and Bottomley PJ, Nitrogen mineralization and immobilization, in
524 *Nitrogen in Agricultural Systems*, ed. by Schepers JS and Raun WR. Soil Science
525 Society of America, Madison, pp. 157-172 (2008).

	-	Varial	ble level*		IIIIO	IY (%) [‡]		
Assay	X_1	X_2	X_3	X_4	 HUC - activity[†] 	Exptl	Predicted	
1	4	6	6	30	730	85	80	
2	4	2	6	180	2174	93	92	
3	4	2	10	30	2347	101	101	
4	4	2	10	180	2055	88	87	
5	4	4	8	105	1326	110	114	
6	4	2	6	30	1986	85	83	
7	4	6	6	180	780	109	114	
8	4	6	10	30	825	98	102	
9	4	6	10	182	961	116	114	
10	27	2	8	105	1402	59	67	
11	27	6	8	105	645	74	71	
12	27	4	8	180	1139	93	93	
13	27	4	6	105	606	78	85	
14	27	4	8	30	1072	88	95	
15	27	4	10	105	971	78	78	
16	50	6	6	30	692	80	84	
17	50	2	6	180	2100	90	89	
18	50	2	10	30	2141	92	90	
19	50	2	10	180	1270	53	53	
20	50	4	8	105	1167	96	99	
21	50	2	6	30	2450	106	103	
22	50	6	6	180	835	99	95	
23	50	6	10	30	678	78	75	
24	50	6	10	180	539	59	64	
25 (cp)	27	4	8	105	1123	92	87	
26 (cp)	27	4	8	105	1106	90	87	
27 (cp)	27	4	8	105	1199	99	87	

Table 1. Experimental design and results according to the central composite facecentered design.

*Variable level X_1 , X_2 , X_3 and X_4 were temperature (°C), HA concentration (mg mL⁻¹), immobilization pH and contact time, respectively.

[†]In U mg⁻¹ HA [‡]IY = Immobilization yield. Soil HA solution: $A_{\rm H} = 99$ U mg⁻¹ HA; urease activity added for immobilization: $A_0 = 4453 \text{ U}$

527

529

Table 2. Urea	ase ac	tivity	/ and	$\mathrm{NH_4^+}$ -	N of soi	l pla	inted with	non	-treat	ted see	ds (con	trol)	and se	eds
encapsulated	with	FU	and	HUC.	Values	are	presented	as	the	mean	(n=3)	and	SEM	(in
parenthesis)*														

		Seed treatment							
Parameter	DAP	Control	FU	HUC	SED (4DF) [†]	SED (4 DF) [‡]	SED (4 DF) [§]		
Urease activity	7	178 (28)	176 (36)	147 (13)	ns	ns	ns		
(U g ⁻¹ dried soil)	14	148 (18)	156 (26)	176 (11)	ns	ns	ns		
	21	137 (4)	148 (5)	150 (9)	ns	ns	ns		
	28	154 (0)	142 (1)	158 (7)	1.1	ns	1.1		
	35	156 (6)	180 (6)	189 (4)	8.6	6.9	ns		
NH4 ⁺ -N	7	2.7 (0.11)	2.9 (0.18)	2.9 (0.12)	ns	ns	ns		
$(mg NH_4^+-N kg^{-1} dried$	14	3.2 (0.01)	2.7 (0.12)	3.3 (0.06)	0.1	ns	0.1		
S011)	21	3.4 (0.11)	5.3 (0.27)	4.1 (0.22)	0.3	0.2	ns		
	28	7.3 (0.29)	9.2 (0.28)	12.0 (0.01)	0.4	0.3	0.3		
	35	2.6 (0.25)	2.8 (0.13)	3.2 (0.06)	ns	0.3	0.1		

* Soil urease activity at t=0: 132 (7) U g⁻¹ dried soil. NH_4^+ -N at t=0: 2.5 (0.06) mg NH_4^+ -N kg⁻¹ dried soil.

[†] SED to compare the coated seeds with FU and control at different DAP.

 \ddagger SED to compare the coated seeds with HUC and control at different DAP.

\$ SED to compare the coated seeds with FU and HUC at different DAP.

ns, not significant; DF, degrees of freedom.

1 /			S	and treatmont	ŀ				
Parameter	DAP	Seed freatment SED SED S Control FU HUC (8 DF)* (8 DF)† (8							
Shoot length	7	93 (5.5)	129 (4.4)	128 (4.2)	7.0	6.9	ns		
(mm)	14	203 (8.4)	265 (9.4)	244 (5.8)	12.6	10.2	9.4		
	21	261 (16.1)	306 (12.0)	314 (13.9)	16.2	17.7	ns		
	28	291 (17.8)	328 (8.7)	321 (6.6)	13.2	11.9	ns		
	35	316 (11.5)	319 (7.5)	314 (7.7)	ns	ns	ns		
Shoot dry weight	7	13.1 (0.67)	18.2 (0.77)	18.1 (0.57)	1.0	0.9	ns		
(mg plant ⁻¹)	14	27.6 (1.02)	43.6 (2.20)	34.5 (1.84)	2.4	2.1	2.2		
	21	49.0 (2.26)	63.6 (4.56)	63.1 (3.45)	5.1	4.1	ns		
	28	67.1 (4.05)	72.3 (2.41)	72.0 (2.66)	ns	ns	ns		
	35	88.1 (1.55)	101.4 (4.25)	92.6 (3.43)	4.5	ns	4.3		
Shoot N	7	42.6 (1.55)	64.4 (2.39)	49.5 (1.07)	2.8	1.9	2.4		
(g kg ⁻¹ dried weight)	14	22.8 (0.57)	30.0 (0.57)	30.8 (0.63)	0.8	0.8	ns		
	21	23.6 (0.76)	30.6 (0.49)	36.7 (1.12)	0.9	1.4	0.5		

23.8 (0.92)

11.6 (0.31)

1.15 (0.043)

1.32 (0.025)

1.99 (0.032)

1.67 (0.064)

1.24 (0.034)

26.9 (0.52)

14.0 (0.33)

0.88 (0.016)

1.04 (0.021)

2.38 (0.073)

1.98 (0.039)

1.26 (0.030)

1.2

0.5

0.05

0.03

0.05

0.08

0.05

0.9

0.5

0.03

0.03

0.08

0.06

0.04

0.9

0.3

0.04

0.03

0.03

0.06

ns

Table 3. Influence of the seed encapsulation with FU and HUC on the plant growth (dry weight, length and shoot N content) and N uptake. Values are presented as the mean (n=5) and SEM (in parenthesis)

* SED to compare the coated seeds with FU and control at different DAP.

[†] SED to compare the coated seeds with HUC and control at different DAP.

20.0 (0.72)

7.5 (0.40)

0.56 (0.020)

0.63 (0.016)

1.17 (0.038)

1.33 (0.048)

0.62 (0.033)

[‡] SED to compare the coated seeds with FU and HUC at different DAP.

ns, not significant; DF, degrees of freedom.

28

35

7

14

21

28

35

545

N uptake

(mg plant⁻¹)

546

547

549	FIGURE CAPTIONS
550	
551	Figure 1. Response surface plot for immobilization yields as a function of: a)
552	temperature and contact time (HA concentration, 4 mg mL ⁻¹ ; pH, 8); b) immobilization
553	pH and HA concentration (temperature, 4 °C; contact time, 180 min).
554	
555	Figure 2. Lineweaver-Burk plot for free (FU) and immobilized (HUC) urease activity.
556	
557	Figure 3. Effect of pH (a) and temperature (b) on free (FU) and immobilized (HUC)
558	urease activity.
559	
560	Figure 4. Stability in soil of free (FU) and immobilized (HUC) urease.
561	
562	
563	
564	
565	
566	
567	
568	
569	
570	
571	
572	





(a)

Figure 1. Response surface plot for immobilization yields as a function of: a)
temperature and contact time (HA concentration, 4 mg mL⁻¹; pH, 8); b) immobilization
pH and HA concentration (temperature, 4 °C; contact time, 180 min).





581 Figure 2. Lineweaver-Burk plot for free (FU) and immobilized (HUC) urease activity.



590 Figure 3. Effect of pH (a) and temperature (b) on free (FU) and immobilized (HUC)591 urease activity.





606 Figure 4. Stability in soil of free (FU) and immobilized (HUC) urease.